1	RECORD OF ORAL HEARING
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3	UNITED STATES PATENT AND TRADEMARK OFFICE
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6	BEFORE THE BOARD OF PATENT APPEALS
7	AND INTERFERENCES
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10	Ex parte YASUO MATSUMURA,
11	KAZUHIKO YANAGIDA, MAILED
12	MANABU SERIZAWA,
13	HIDEKAZU YAGUCHI, JUL 0 3 2007
14	TSUTOMU KUBO, PAT. & T.M. OFFICE
15	and SHIGERU SEITOKU  BOARD OF PATENT APPEALS AND INTERFERENCES
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18	Appeal 2007-1025
19	Application 10/658,811
20	Technology Center 1700
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23	Oral Hearing Held: May 9, 2007
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27	Before CATHERINE Q. TIMM, JEFFREY T. SMITH, and [FULL LINDA
28	M. GAUDETTE,
29	Administrative Patent Judges
30	
31	On Behalf of the Appellant:
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33	RYAN CHRISTOPHER CADY, ESQ.
34	Oliff & Berridge, PLC
35	P.O. Box 19925
36	Alexandria, Virginia 22320
37	(703) 836-6400
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1	The above-entitled matter came on for hearing on Wednesday,
2	May 9, 2007, commencing at 9:45 a.m. at the U.S. Patent and Trademark
3	Office, 600 Dulany Street, 9 <sup>th</sup> Floor, Alexandria, Virginia.
4	CLERK: Calendar No. 18, Case No. 2007-1025. Mr. Ryan
5	Cady.
6	(Mr. Cady entering the hearing room.)
7	JUDGE TIMM: Good morning, Mr. Cady.
8	MR. CADY: Good morning, Your Honors.
9	JUDGE TIMM: Could you spell your last name for the record,
10	please?
11	MR. CADY: Yes. C-a-d-y.
12	JUDGE TIMM: And you have 20 minutes to present your
13	arguments. And you can assume that we understand the issues and the facts
14	of the case. And you may start when you're ready.
15	MR. CADY: Thank you very much. On appeal in this case are
16	essentially 102(b) and 103 rejections over the primary reference, which is
17	Tanaka. The application generally relates to improved toner compositions
18	and binder resins, which are well-known in the art.
19	And Tanaka has essentially been used to anticipate claim 1,
20	from which all other claims depend. And for that reason, we'd like to focus

- the discussion today entirely on the 102(b) rejection over claim 1, using
- 2 Tanaka. The facts surrounding the Tanaka reference are so clear that we
- 3 believe that we can overcome all of the obvious rejections as well.
- 4 Essentially, in this case the examiner failed to establish the
- 5 prima facie case that Tanaka discloses the features of claim 1. The
- 6 examiner points to Tanaka's claim 11, which recites a co-polymer that is a
- 7 methyl methacrylate, isobutyl methacrylate, and methacrylic acid
- 8 co-polymer. That's true.
- 9 The problem with that analysis is that the isobutyl methacrylate
- in Tanaka's claim 11 and the butyl methacrylate which we require, which
- meet the limitations of our instant claim 1, are a different species. Claim 1
- requires the butyl methacrylate because it has a glass transition point below
- 13 50 degrees, whereas the examiner assumes that the isobutyl methacrylate
- has the same glass transition point.
- But it doesn't. In actuality, the isobutyl form has a glass
- transition of over 50 degrees. And so in that respect, the examiner clearly
- failed in his -- to bear the initial burden of establishing a prima facie case
- 18 for anticipation.

1	JUDGE SMITH: As a point of clarity, when you refer to Tg
2	monomer, you're meaning a polymer produced from that monomer having
3	that Tg point?
4	MR. CADY: If I can refer to our claim, the claim
5	specifically the way it's stated is that the toner comprises as a main
6	component of binder resin, that the binder resin has a copolymer consisting
7	of three monomers, so the monomer 1, 2, and 3. The first monomer is the
8	high Tg monomer.
9	JUDGE SMITH: Monomers don't have transition glass
10	temperatures, usually. That's why I'm asking for a point of clarification.
11	MR. CADY: Without well, I won't make any statements
12	about broad chemistry in general. But certainly here you can we've even
13	in fact done a search result for these monomers. And I'd like to talk about
14	that in a second. But in this case, the monomers in our specification that we
15	recite and that are also recited in Tanaka claim 11 each have or the first
16	two, at least have a glass transition temperature.
17	JUDGE SMITH: What's the molecular weight of your
18	monomer?
19	MR. CADY: I believe it's in the range that the examiner cites,
20	which is around I'd have to read the specification. I'm not prepared to

answer exactly what the weight is, although we don't traverse the examiner's 1 assertion that it does have a -- his assumed molecular weight in his answer. 2 3 If I may? JUDGE SMITH: So what evidence do you have that the 4 5 monomers listed in the claim 11 of Tanaka don't meet your claim, then? Didn't he produce that on the record? 6 MR. CADY: It's not clear we traverse the examiner's assertion 7 that he's met the prima facie case. We didn't plainly and simply say that the 8 9 isobutyl form has a different glass transition temperature. I'm not sure why. I also don't know why the examiner didn't take the .8 seconds that it 10 11 required for us to do the search to make the same determination. 12 We've got those here for your convenience, if you're interested at all. There are over 100,000 results. Literally the first document that we 13 14 pulled has a glass transition temperature of both the isobutyl methacrylate 15 and the normal butyl methacrylate. 16 JUDGE TIMM: Well, if that evidence wasn't before the examiner, I don't think it would be appropriate for us to look at it now. 17 18 MR. CADY: That's fine. Would the board be interested in 19 taking judicial notice of that fact, being that it's --20 JUDGE SMITH: No.

1	MR. CADY: readily available and widely accepted?
2	JUDGE SMITH: No.
3	JUDGE TIMM: No. We would like to be able to have the
4	examiner's input before we would make a determination like that since it
5	wasn't discussed in the record.
6	MR. CADY: Okay. That being said, that's essentially what the
7	issue here is, as we see it, is that the examiner failed in that respect to do the
8	requisite work in order to make to establish the prima facie case of
9	anticipation because in fact, one of the monomers listed in Tanaka's claim
10	11 simply doesn't meet the limitations of our claim 1. And having overcome
11	Tanaka, we then
12	JUDGE SMITH: Okay. Let's back up for a second just to
13	clarify your record. You're saying the isobutyl methacrylate has a Tg of
14	what?
15	MR. CADY: Fifty-three.
16	JUDGE SMITH: And the other monomer okay. For that
17	matter, give us all three Tgs. The methyl methacrylate is?
18	MR. CADY: Methyl methacrylate? I don't have I've got
19	isobutyl methacrylate and butyl methacrylate. We don't contest that methyl

methacrylate and the hydrophilic monomer that the examiner cites are 1 within the rages. We accept that. 2 3 JUDGE SMITH: What's your Tg for butyl methacrylate? 4 MR. CADY: Twenty. So it's beneath 50 degrees, and it 5 qualifies in our specification and in our claim as a low Tg monomer. 6 JUDGE SMITH: But according to claim 11, you have a methyl methacrylate, an isobutyl methacrylate, and a methacrylic acid copolymer. 7 8 MR. CADY: Claim 11 of Tanaka, Your Honor? JUDGE SMITH: Yes. Yes. 9 10 MR. CADY: That's right. And the --11 JUDGE SMITH: And then according to paragraph 40 of your specification -- let's see. 12 13 MR. CADY: That's exactly right. JUDGE SMITH: So you're saying both methyl methacrylate 14 and the isobutyl methacrylate both were high Tg monomers? 15 16 MR. CADY: Yes. The examiner uses -- if you look at claim 17 11, the middle monomer there is the isobutyl methacrylate. That, according 18 to our claim limitations, needs to be a low Tg monomer. The examiner looks at the examples of our low Tg monomers and equates the isobutyl 19

form in Tanaka with the butyl methacrylate that we list, which is -- those are 1 separate species and they have different properties. 2 3 In our specification, you can see that we treat iso -- I think in paragraph 45, there's a specific example of a high Tg monomer that's the iso 4 form. And throughout other parts of the specification, we truly -- we clearly 5 6 treat the iso form as different than the normal form of these monomers. And 7 I think that's also well-established in the art. 8 JUDGE GAUDETTE: But you never raised that argument during prosecution? 9 10 MR. CADY: It doesn't seem that it was raised specifically. It 11 seems that they simply argued that the prima facie case hadn't been met. I 12 don't know why they didn't take the extremely brief amount of time it takes to determine that, or why the examiner didn't meet his burden in that 13 14 respect. JUDGE TIMM: Is there another argument that is on the record 15 16 that you'd like to have us consider? 17 MR. CADY: All the arguments relating to -- certainly relating 18 to obviousness, but also relating to the general thrust of Tanaka having been directed toward a styrenic compound, clearly the examples in Tanaka, both 19 examples and the entire disclosure, teaches away from in all respects using 20

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particular species.

1 our compound, in the sense that throughout our specification teach away from using styrene for all of the detriments in its entirety. And I think those 2 3 arguments are clearly briefed. 4 JUDGE TIMM: But is there an argument that claim 11 in the 5 reference doesn't anticipate? 6 MR. CADY: There is -- well, we don't concede in any sense 7 that claim 11 anticipates. And clearly, with this, what I'm arguing now -- if 8 I understand your question -- what I'm arguing now is simply clearly that claim 11 does not anticipate. 9 JUDGE TIMM: But of your arguments on the record? 10 11 MR. CADY: Our arguments on the record don't concede that 12 claim 11 anticipates. They don't seem to traverse on the same grounds, as far as I can tell. They do only in the general sense that we say that the 13 14 examiner hasn't made out a prima facie case. We don't specifically say that 15 this butyl methacrylate is actually a low Tg monomer whereas the isobutyl 16 form is a higher --17 JUDGE SMITH: In fact, on page 12, first paragraph of your 18 brief, you seem to say that it -- you seem to acknowledge that it does meet 19 your claim, but argue further that there are no examples shown in this

1	MR. CADY: We say that Tanaka's claim 11 indicates, and we
2	there I guess we recite exactly what Tanaka recites. But we would argue
3	that that's not a concession. That's simply a recitation of what Tanaka
4	actually recites. We know, and ordinarily skilled artisans know, that the
5	isobutyl form is different from the butyl methacrylate.
6	JUDGE SMITH: But how does that correspond to the
7	following, taken from that same paragraph: "Nor does Tanaka indicate that
8	these monomers are so chosen that a high Tg monomer, a low Tg monomer,
9	and a hydrophilic monomer are included in the copolymer"?
10	MR. CADY: Well, I think that's it's got to be a logical
11	conclusion because certainly the isobutyl form isn't a low Tg monomer.
12	And so Tanaka couldn't indicate that the monomers were chosen for those
13	reasons.
14	JUDGE SMITH: Okay. You can go on.
15	MR. CADY: Those are primarily the arguments that we'd like
16	to bring up, in addition to everything that's been briefed, as to the other
17	the general notion that Tanaka teaches directly away from our claimed
18	invention, that it's examples and its entire disclosure is directed towards
19	styrene-containing compounds.

1	Essentially well, in addition to that, you would say that an
2	ordinarily skilled artisan looking at Tanaka, being that it's entirely directed
3	towards styrene-containing compounds, wouldn't have been motivated or
4	there would be no suggestion to have combined Tanaka with any of the
5	secondary references in order to obtained the claimed features.
6	And having said that, we don't believe that Tanaka alone does.
7	And so, essentially, all the obviousness rejections would fall as well.
8	JUDGE TIMM: Do you have any other arguments?
9	MR. CADY: No, Your Honor, I don't. If I may ask again that
10	the board take official notice of the fact that the isobutyl form has a different
11	glass transition than the butyl form of methacrylate. And that's all I've got.
12	JUDGE TIMM: Okay. I think we understand your position.
13	Judge Gaudette, Judge Smith, do you have any more questions?
14	JUDGE SMITH: No.
15	JUDGE GAUDETTE: No.
16	MR. CADY: Very good. Thank you.
17	(Whereupon, at 9:58 a.m., the appeal was concluded.)